Appendix F Radionuclides Detected by Gamma Spectroscopy (Gamma Scan)

One of the several forms of radiation is gamma radiation. Gamma radiation is emitted by many radionuclides. Gamma spectroscopy, sometimes called a gamma scan, is used to detect the presence of the radionuclides shown in Table F.1. These radionuclides may be natural or result from Hanford Site operations. They include activation products formed by the absorption of a neutron by a stable element and fission products that occur following fission (splitting) of nuclear fuel radionuclides such as uranium-235 or plutonium-239. Some of these radionuclides may not be discussed in the main body of this report if they are below detection levels.

<u>Radionuclide</u>	<u>Symbol</u>	<u>Source</u>
Beryllium-7 ^(a)	⁷ Be	Natural
Sodium-22	²² Na	Activation product
Sodium-24	²⁴ Na	Activation product
Potassium-40 ^(a)	⁴⁰ K	Natural
Manganese-54	⁵⁴ Mn	Activation product
Cobalt-58	⁵⁸ Co	Activation product
Cobalt-60 ^(a)	⁶⁰ Co	Activation product
Iron-59	⁵⁹ Fe	Activation product
Zinc-65	⁶⁵ Zn	Activation product
Zirconium/niobium-95	⁹⁵ Zr/Nb	Activation product and fission produc
Molybdenum-99	⁹⁹ Mo	Activation product and fission produc
Ruthenium-103	¹⁰³ Ru	Activation product and fission produc
Ruthenium-106 ^(a)	¹⁰⁶ Ru	Fission product
Antimony-125 ^(a)	¹²⁵ Sb	Activation product
Iodine-131	¹³¹ I	Fission product
Cesium-134 ^(a)	^{134}Cs	Activation product
Cesium-137 ^(a)	¹³⁷ Cs	Fission product
Barium/lanthanum-140	¹⁴⁰ Ba/La	Fission product
Cerium-141	¹⁴¹ Ce	Activation product and fission produc
Cerium/praseodymium-144	¹⁴⁴ Ce/Pr	Fission product
Europium-152	¹⁵² Eu	Activation product
Europium-154 ^(a)	¹⁵⁴ Eu	Activation product
Europium-155 ^(a)	¹⁵⁵ Eu	Activation product